

9.12) For H_2 at 293K a) FIND v_{mp} , THEN USE THE APPROXIMATION IN EXAMPLE 9.4 TO DETERMINE THE FRACTION OF MOLECULES IN THE RANGES 0.9 TO 1.1.

b) $0.8 v_{mp}$ TO $1.2 v_{mp}$, c) $0.5 - 0.6 v_{mp}$ AND d) $1.5 - 1.6 v_{mp}$

a) For a CLASSICAL GAS AND $M_{H_2} = 2(1.008 u)(1.66 \times 10^{-27}) = 3.35 \times 10^{-27} \text{ kg}$

$$v_{mp} = \sqrt{\frac{2kT}{m}} = \sqrt{\frac{2(1.38 \times 10^{-23})(293)}{3.35 \times 10^{-27}}} = 1553.7 \frac{m}{s}$$

b) - d) USE THE APPROXIMATION

$$P = F(v) \Big|_{v_0}^{\Delta v} = 4N \left(\frac{m}{2\pi kT} \right)^{3/2} e^{-\frac{1}{2} \left(\frac{mv^2}{kT} \right)} v^2 \Big|_{v_0}^{\Delta v}$$

EVALUATE FACTORS:

$$4N \left(\frac{m}{2\pi kT} \right)^{3/2} = 4N \left(\frac{3.35 \times 10^{-27}}{2\pi(1.38 \times 10^{-23})(293)} \right)^{3/2} = 4N (4.79 \times 10^{-11})$$

$$= 6.02 \times 10^{-10} \frac{s^3}{m^3}$$

$$\left(\frac{m}{2kT} \right) = \frac{3.35 \times 10^{-27}}{4(1.38 \times 10^{-23})(293)} = 4.14 \times 10^{-7} \frac{s^2}{m^2}$$

$$\Rightarrow P = (6.01 \times 10^{-10}) e^{-(4.14 \times 10^{-7}) v^2} v^2 \Big|_{v_0}^{\Delta v}$$

THEN FOR $v_0 =$ THE CENTER OF EACH SPAN

b) $v_0 = v_{mp}$, $\Delta v = 0.1 v_{mp}$

$$P_{0.95-1.05 v_{mp}} = (6.01 \times 10^{-10}) e^{-(4.14 \times 10^{-7})(1553.7)^2} (1553.7)^2 (155.4)$$

$$P_{0.95-1.05 v_{mp}} = 0.0830 \Rightarrow 8.3\% \text{ OF PARTICLES IN THIS SPAN}$$

c) $v_0 = 0.55 v_{mp}$, $\Delta v = 0.1 v_{mp}$

$$P_{0.5-0.6 v_{mp}} = (6.01 \times 10^{-10}) e^{-(4.14 \times 10^{-7})(854.5)^2} (854.5)^2 (155.4)$$

$$P_{0.5-0.6 v_{mp}} = 0.0504 \Rightarrow 5.04\% \text{ OF PARTICLES IN THIS SPAN}$$

d) $v_0 = 1.55 v_{mp}$, $\Delta v = 0.1 v_{mp}$

$$P_{1.5-1.6 v_{mp}} = (6.01 \times 10^{-10}) e^{-(4.14 \times 10^{-7})(2408)^2} (2408)^2 (155.4)$$

$$P_{1.5-1.6 v_{mp}} = 0.0491 \Rightarrow 4.9\% \text{ OF PARTICLES IN THIS SPAN}$$