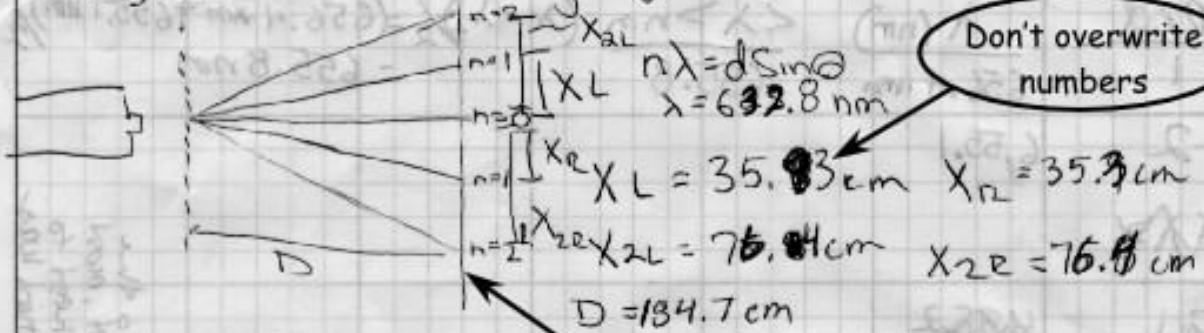
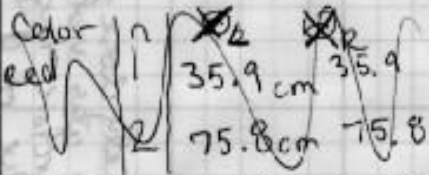


Larry, Moe, Curly Joe, Diffraction Grating and the Hydrogen Spectrum July 5, 2010



Don't overwrite numbers

Sketch too small, messy



$\theta_1 = \tan^{-1}\left(\frac{X_L}{D}\right) = \tan^{-1}\left(\frac{35.3 \text{ cm}}{134.7 \text{ cm}}\right) = 14.82^\circ$

$\theta_2 = \tan^{-1}\left(\frac{X_{2L}}{D}\right) = \tan^{-1}\left(\frac{76.4 \text{ cm}}{134.7 \text{ cm}}\right) = 29.7^\circ$

Tough to follow!

$n=1 \quad d = \frac{n\lambda}{\sin\theta} = \frac{1(632.8 \times 10^{-9} \text{ m})}{\sin(14.82^\circ)} = 3.371 \times 10^{-6} \text{ m}$

$n=2 \quad d = 2(632.8 \times 10^{-9} \text{ m}) = 3.311 \times 10^{-6} \text{ m}$

n=1

Need more room for table

Order	Color	θ_R (deg, min)	θ_L (deg, min)	θ_R (decimal)	θ_L (decimal)	θ_R	θ_L	θ
1	Red	116°34'	139°14'	116.57°	139.23°	11.35°	-11.31°	11.33°
1	Blue-Green	119°34'	136°16'	119.57°	136.27°	8.35°	-8.35°	8.35°
1	Violet	120°27'	135°20'	120.45°	135.33°	7.47°	-7.41°	7.44°

Difficult to tell which calculations produce which results

$\langle d \rangle = \frac{(3.371 \times 10^{-6} \text{ m} + 3.311 \times 10^{-6} \text{ m})}{2} = 3.341 \times 10^{-6} \text{ m}$

n	Color	θ_R (deg, min)	θ_L (deg, min)	θ_R (decimal)	θ_L (decimal)	θ_R	θ_L	θ
2	Red	151°03'	104°53'	151.05°	104.88°	23.04°	-23.13°	23.09°
4	Red	92°28'	143°30'	92.47°	143.50°	35.45°	-35.58°	35.52°

Leave room for corrections

2nd order Violet could not be seen

$$\lambda = d \sin \theta \quad \text{Red}(n=1) = (3.341 \times 10^{-6} \text{ m}) (\sin 11.33^\circ) = 6.564 \times 10^{-7} \text{ m}$$

Red	λ (nm)	$\langle \lambda \rangle$ nm	$(\lambda_1 + \lambda_2) / 2 = (656.4 \text{ nm} + 655.1 \text{ nm}) / 2 = 655.8 \text{ nm}$
1	656.4 nm	655.8	
2	655.1		

Don't write so close to the margins. Plan ahead.

~~Red~~

Red	λ (nm)	$\langle \lambda \rangle$ nm
1	485.2	485.3
1	485.2	
2	485.3	

Cross out with a single line

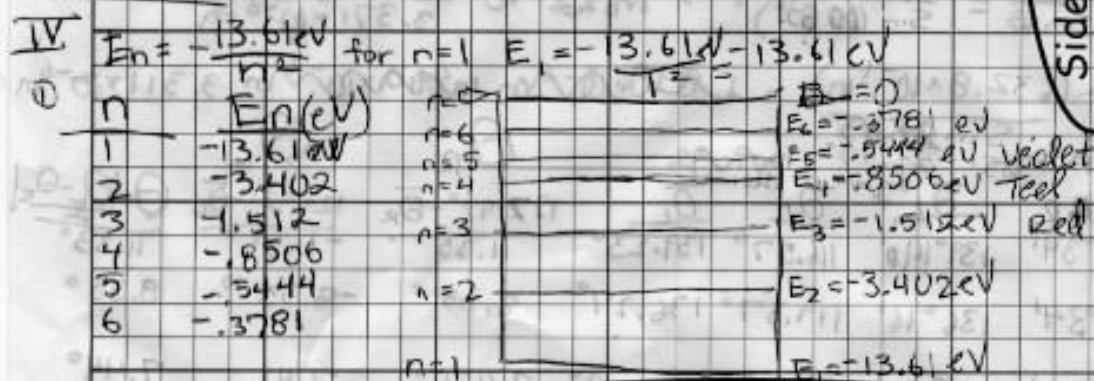
Sideways?! Use another sheet of paper!

Create a proper table, so that it's easy to tell what columns a value belongs with.

How close? Quote percent difference!

~~Red~~

Violet	λ (nm)	$\langle \lambda \rangle$ nm
1	432.6	432.6



Sketch too small, and incomplete.

② n=6,5,4,3 to n=2

$$E_{ph} = E_i - E_f =$$

6 → 2 = -0.3781 eV - -3.402 eV = 3.024 eV

5 → 2 = 2.857 eV

4 → 2 = 2.551 eV

3 → 2 = 1.89 eV

DISCUSSION
The measured values of the experiment were extremely close to the expected values. The % error that was the smallest was violet, most likely because it was the toughest to see and therefore brightest to measure. Uncertainty can be attributed to the value of d that was determined due to uncertainty of the measurements.

Begin discussion by restating the numerical results.

$\lambda = \frac{hc}{E_{ph}}$	Wavelength (nm)	% Error
6 → 2	1239.9 eV·nm / 3.024 eV = 410.0 nm	
5 → 2	433.434.0 nm	% Error = 434.0 - 432.6 / nm × 100 = .32% Violet
4 → 2	486.0 nm	
3 → 2	656.0 nm	

Blue-Green = .14%
Red = .03%

Use a leading zero. e.g. 0.32%